CAMBRIDGE INTERNATIONAL EXAMINATIONS International General Certificate of Secondary Education

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0654 CO-ORDINATED SCIENCES

0654/61

Paper 6 (Alternative to Practical), maximum raw mark 60

This mark scheme is published as an aid to teachers and candidates, to indicate the requirements of the examination. It shows the basis on which Examiners were instructed to award marks. It does not indicate the details of the discussions that took place at an Examiners' meeting before marking began, which would have considered the acceptability of alternative answers.

Mark schemes should be read in conjunction with the question paper and the Principal Examiner Report for Teachers.

Cambridge will not enter into discussions about these mark schemes.

Cambridge is publishing the mark schemes for the October/November 2012 series for most IGCSE, GCE Advanced Level and Advanced Subsidiary Level components and some Ordinary Level components.

Page		Syllabus
	IGCSE – October/November 2012	0654 230
(a) (i)	<i>first row</i> : 10, 10, 10, 10 ;	anth
(ii)	second row: 0, 9, 0, 10 ;	Syllabus 0654 PathaCambrid
	ht not necessary ; iter is necessary ;	[2]
	improve reliability/because some seeds might be dead or c count of individual variability ;	damaged/to take [1]
ter	y two of: nperature ; ygen/air ;	
рH	;	[max 2]
• •	arch – seeds ; ducing sugar – radicles/roots ;	[2]
(f) am	nylase/carbohydrase/diastase ;	[1]
• •		[Total: 10]
(a) (i)	64.5 ;	[0]
	59.2 ;	[2]
	(64.5 – 40 =) 24.5 and (59.2 – 40 =) 19.2 (both correct) ;	[1]
(iii)	1/70 = 0.014 ; 1/90 = 0.011 ; (penalise incorrect d.p. once only)	[2]
(b) (i)	correct plots of 4 or 5 points ; straight line drawn ;	[2]
(ii)	<i>x</i> - and <i>y</i> - distances shown on graph ; <i>y</i> / <i>x</i> correctly calculated (1600 to 1800) ;	[2]
	0 – gradient/10 correctly calculated from candidate's graph	
14	0), do not allow impossible masses e.g. negative ;	[1]

Page 3	Mark Scheme	Syllabus 🔪 😪 🗽
	IGCSE – October/November 2012	0654
(a) same m	ass of soil/same volume of water ;	Syllabus 0654 Anne Anne Anne Anne Anne Anne Anne Anne
(b) (from) bl (to) red ;		
(c) (i) 4.4 4.9 5.2		
(ii) 5.6,	5.1, 4.8 (all three, ecf) ;	
(iii) (5.6	+ 5.1 + 4.8 = 15.5, 15.5/3 =) 5.17 OR 5.2 ;	
	3 × 10/5.2 = 0.05 (mol/dm ³) (ecf) ; nore d.p.)	
(e) the (inso	luble) <u>hydroxides</u> (of the metals) are formed/owtte ;	
		[Total:

 condition of leaves	time/ mins	reading on scale/ cm	distance moved by bubble per minute/cm	average distance moved by bubble per minute/cm
	1	1.6	1.6	1.57
untreated	2	3.3	1.7	OR 1.6
	3	4.7	1.4	

(i)	as in column 3 ;	[1]
(ii)	as in column 4 ;	[1]
(iii)	as in column 5 ;	[1]
(b) (i)	1.2/1.6 × 100 ; = 75 % ; (accept 76 % if 1.57 used)	[2]
(ii)	cover the lower surface with grease (this should stop all transpiration);	

 (ii) cover the lower surface with grease (this should stop all transpiration);
(candidates may suggest to repeat the experiment, this time with untreated and then lower surface greased. the mark should be allowed for this)

[1]

Page 4	Mark Scheme	Syllabus	. Y.
	IGCSE – October/November 2012	0654	SC.
(c) any two o change in temperat humidity light ;	n air speed ; ture ;	Syllabus 0654	embhing [max 2]
(d) (i) to p	revent air bubbles from entering the shoot ;		[1]
	er used in plant for photosynthesis/mai ansion/produced by respiration ;	ntaining cell turgor/cell	[1]
		l	Total: 10]
(a) 1a green 1b purple			[2]
(b) (sodium)	sulfate ;		[1]
(c) (sodium) (sodium)			[2]
(litr	mus is blue at first and then) turns red ; mus is blue at first and then) turns red ; bbles are given off ;		[3]
(e) (i) bariu	um sulfate;		[1]
(ii) a so	lid is formed from a solution/insoluble solid for	ms ;	[1]
			Total: 10]
(a) (i) heat light	; ; (either order)		[2]
(ii) argo	on OR inert gas ;		[1]
(b) A and V	shown in correct places in the circuit ;		[1]

Page 5	Mark Sche	eme	Syllabus	·
	IGCSE – October/No	vember 2012	0654	M. Papac
(d) (i) 150	/240 = 0.6(25) A ;			f the
• •	resistance must be much hi	gher at the higher	e.m.f. (because o	f the
ngi	ner temperature) ;			
(e) heat is r	nade (instead of light) ;			
(e) heat is r and one	nade (instead of light) ; e of:	ot needed/lost ·		
(e) heat is r and one so that (nade (instead of light) ; e of: electrical) energy is wasted/nergy needs to be generated		to be used (to r	